

GLOBAL MANUFACTURING

ARAUJO, Anna Carla Sept, 2015 Mechanical Engineering Department – POLI/COPPE/UFRJ



Joining Processes [#11]



Joining x Assembly

- **Joining**: is generally used for welding, brazing, soldering, and adhesive bonding, which form a permanent joint between the parts—a joint that cannot easily be separated.
- Assembly: usually refers to mechanical methods of fastening parts together. Some of these methods allow for easy disassembly, while others do not.
- The metallurgical processes of welding, brazing, and soldering are usually used to join metals and often involve the solidification of molten material.
- The use of discrete fasteners (such as bolts and nuts, screws, and rivets) requires the creation of aligned holes and produces stress localization. While the holes may affect performance, disassembly and reassembly can often be performed with relative ease.
- The adhesive bonding has grown with new developments in polymeric materials and is being used extensively in automotive and aircraft production. Any material can be joined to any other material, and the low-temperature joining is particularly attractive for composite materials. Production rates are often low, however, because of the time required for the adhesive to develop full strength.
- From a technical viewpoint, powder metallurgy is another consolidation process, because the end product is built up by the joining of a multitude of individual particles.

Welding

Welding is a materials joining process in which two or more parts are **<u>coalesced</u>** at their contacting surfaces by a suitable application of heat and/or pressure.

Many welding processes are accomplished by

- heat alone, with no pressure applied;
- a combination of heat and pressure;
- Or by pressure alone, with no external heat supplied.

In some welding processes a filler material is added to facilitate coalescence.

- (1) fusion welding and
- (2) solid-state welding

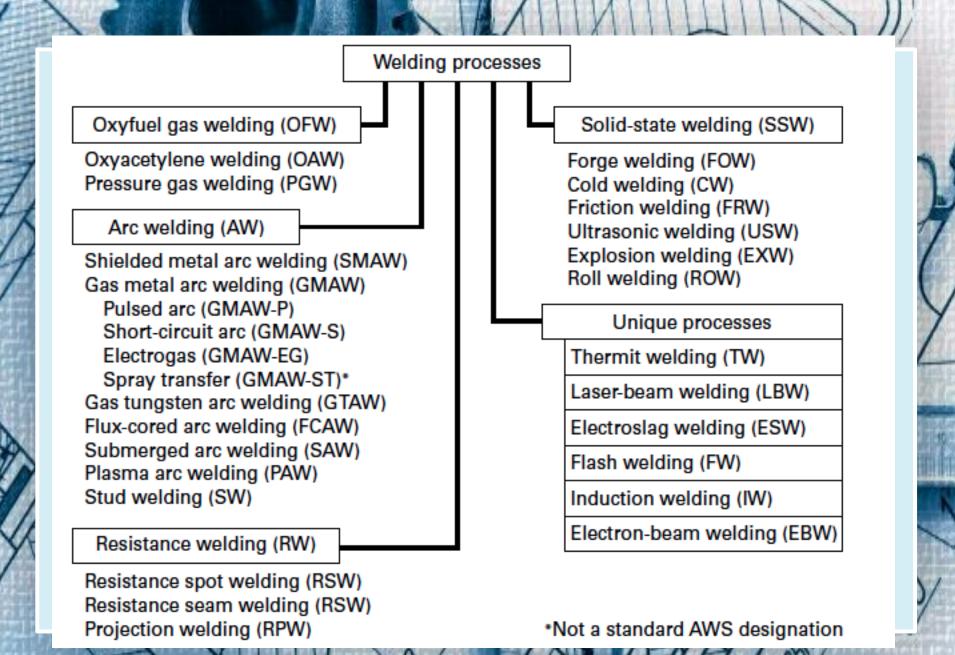
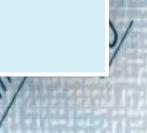


TABLE 30-2 Weldability or Joinability of Various Engineering Materials ^a							
Material	Arc Welding	Oxyacetylene Welding	Electron-Beam Welding	Resistance Welding	Brazing	Soldering	Adhesiv Bonding
Cast iron	С	R	Ν	S	D	Ν	С
Carbon and low-alloy steel	R	R	С	R	R	D	С
Stainless steel	R	С	С	R	R	С	С
Aluminum and magnesium	С	С	С	С	С	S	R
Copper and copper alloys	С	С	С	С	R	R	С
Nickel and nickel alloys	R	С	С	R	R	С	С
Titanium	С	Ν	С	С	D	S	С
Lead and zinc	С	С	Ν	D	Ν	R	R
Thermoplastics	Heated tool R	Hot gas R	Ν	Induction C	Ν	Ν	С
Thermosets	Ν	Ν	Ν	Ν	Ν	Ν	С
Elastomers	Ν	Ν	Ν	Ν	Ν	Ν	R
Ceramics	Ν	S	С	Ν	Ν	Ν	R
Dissimilar metals	D	D	С	D	D/C	R	R

^a R, recommended (easily performed with excellent results); C, commonly performed; D, difficult; S, seldom used; N, not used.



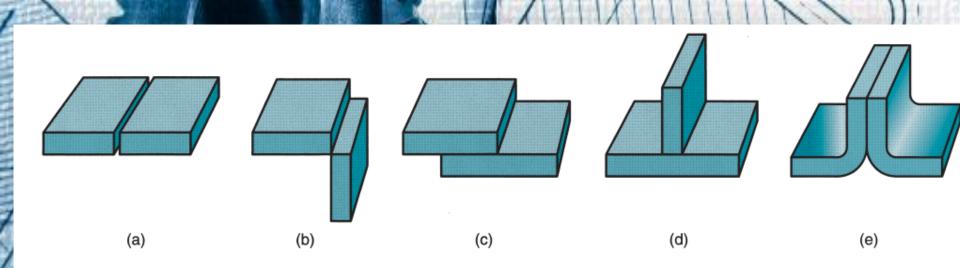
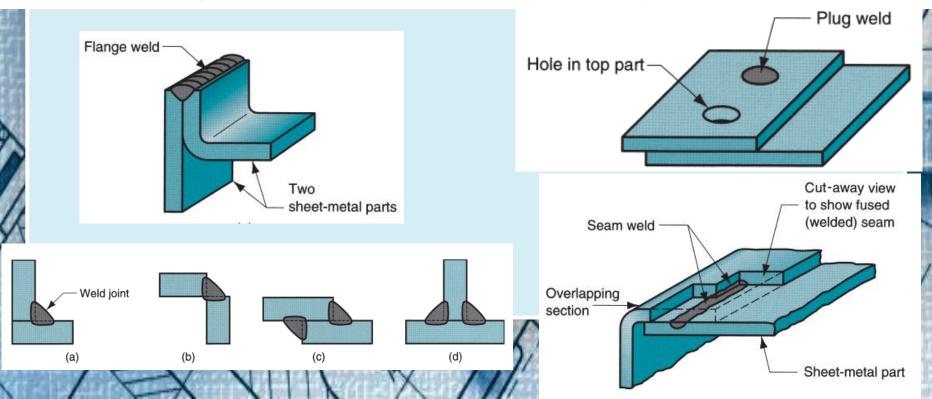


FIGURE 22.2 Five basic types of joints: (a) butt, (b) corner, (c) lap, (d) tee, and (e) edge. (Credit: Fundamentals of Modern Manufacturing, 4th Edition by Mikell P. Groover, 2010. Reprinted with permission of John Wiley & Sons Inc.)



Fusion Welding

- Fusion-welding processes use heat to melt the base metals.
- Filler:
 - A filler metal is added to the molten pool to facilitate the process and provide bulk and strength to the welded joint.
 - A fusion-welding operation in which no filler metal is added is referred to as an autogenous weld.

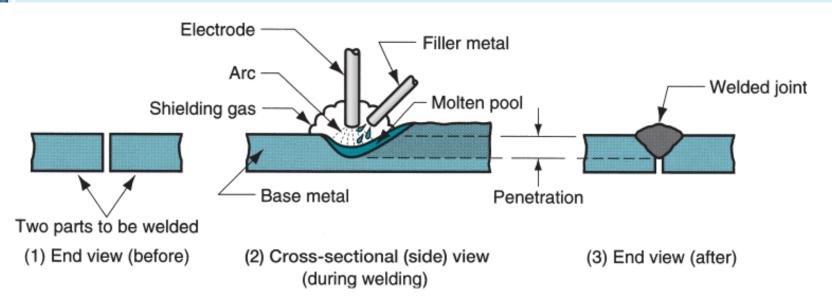


FIGURE 22.1 Basics of arc welding: (1) before the weld; (2) during the weld (the base metal is melted and filler metal is added to the molten pool); and (3) the completed weldment. There are many variations of the arc-welding process. (Credit: *Fundamentals of Modern Manufacturing*, 4th Edition by Mikell P. Groover, 2010. Reprinted with permission of John Wiley & Sons, Inc.)

Physic of Welding – Power Density

Power density can be defined as the **heating power transferred to the work per unit surface area**, W/mm². The time to melt the metal is inversely proportional to the power density.

At low power densities, a significant amount of time is required to cause melting. If power density is too low, the heat is conducted into the work as rapidly as it is added at the surface, and melting never occurs.

PD	=	$\frac{P}{A}$

Differences among welding processes in this range are: (1) the rate at which welding can be performed and/or (2) The size of the region that can be welded.

TABLE 22.1Comparison of several fusion-welding processeson the basis of their power densities.

	Approximate Power Density				
Welding Process	W/mm ²	Btu/sec-in ²			
Oxyfuel welding	10	6			
Arc welding	50	30			
Resistance welding	1000	600			
Laser beam welding	9000	5000			
Electron beam welding	10,000	6000			

Physic of Welding – Heat Balance in Fusion Welding

The quantity of heat required to melt a given volume of metal depends on:

(1) the heat to raise the temperature of the solid metal to its melting point, which depends on the metal's volumetric specific heat,

(2) the melting point of the metal $\mathbf{T}_{\mathbf{m}}$, and

(3) the heat to transform the metal from solid to liquid phase at the melting point, which depends on the metal's heat of fusion.

 $U_m = KT_m^2$

U_m = the unit energy for melting K = Conversion Unit Constant

 $H_w = f_1 f_2 H$

Hw = net HeatF = factors

 $R_{Hw} = f_1 f_2 R_H = U_m A_w v$

TABLE 22.2Melting temperatures on the absolute temperature scale for selected metals.								
	Melting Temperature			Melting Temperature				
Metal	°K ^a	°R ^a	Metal	°K ^a	°R ^b			
Aluminum alloys	930	1680	Steels					
Cast iron	1530	2760	Low carbon	1760	3160			
Copper and alloys			Medium carbon	1700	3060			
Pure	1350	2440	High carbon	1650	2960			
Brass, navy	1160	2090	Low alloy	1700	3060			
Bronze (90 Cu-10 Sn)	1120	2010	Stainless steels					
Inconel	1660	3000	Austenitic	1670	3010			
Magnesium	940	1700	Martensitic	1700	3060			
Nickel	1720	3110	Titanium	2070	3730			

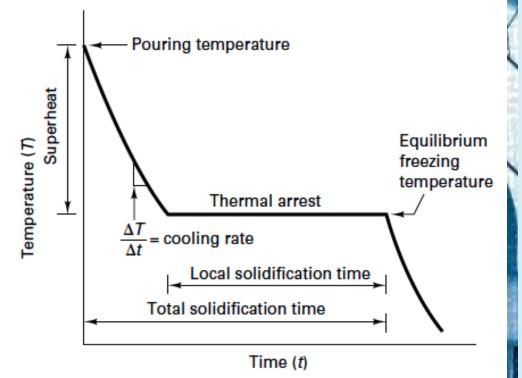
Based on values in [2].

^aKelvin scale = Centigrade (Celsius) temperature + 273.

^bRankine scale = Fahrenheit temperature + 460.

 R_{Hw} = rate of heat energy delivered to the operation for welding

How pure metal solidifies? [CASTING]



Pouring temperature: the temperature of the liquid metal when it first enters the mold.

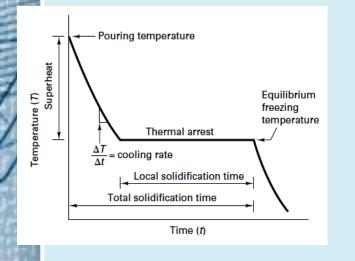
No e TIIN

Superheat: is the difference between the pouring temperature and the freezing temperature of the material.

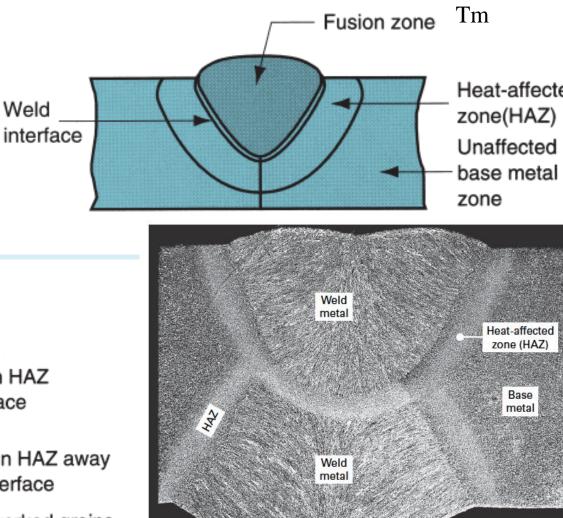
Cooling rate: the rate at which the liquid or solid is cooling and can be viewed as the slope of the cooling curve at any given point.

Thermal arrest: is the plateau in the cooling curve that occurs during the solidification of a material with fixed melting point. At this temperature, the energy or heat being removed from the mold comes from the latent heat of fusion that is being released during the solidification process.

Physic of Welding – Microstructure / Solidification



Columnar grains

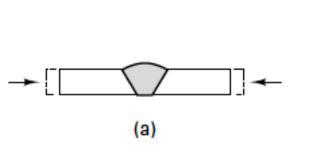


in fusion zone Coarse grains in HAZ

near weld interface

Finer grains in HAZ away from weld interface

Original cold-worked grains



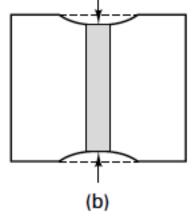


FIGURE 30-15 Shrinkage of a typical butt weld in the transverse (a) and longitudinal (b) directions as the material responds to the induced stresses. Note that restricting transverse motion will place the entire weld in transverse tension.

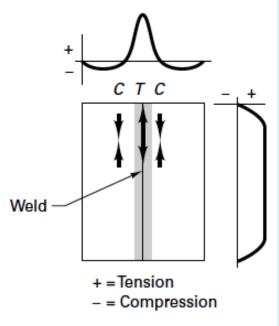
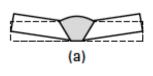
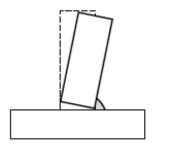
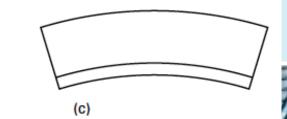


FIGURE 30-14 Schematic of the longitudinal residual stresses in a fusion-welded butt joint.



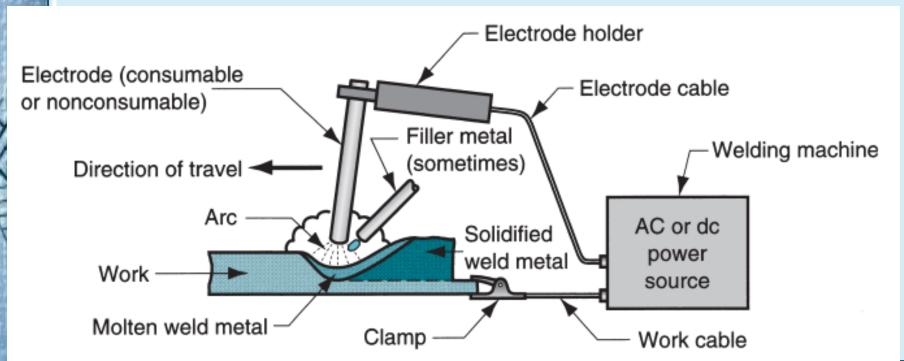


(b)

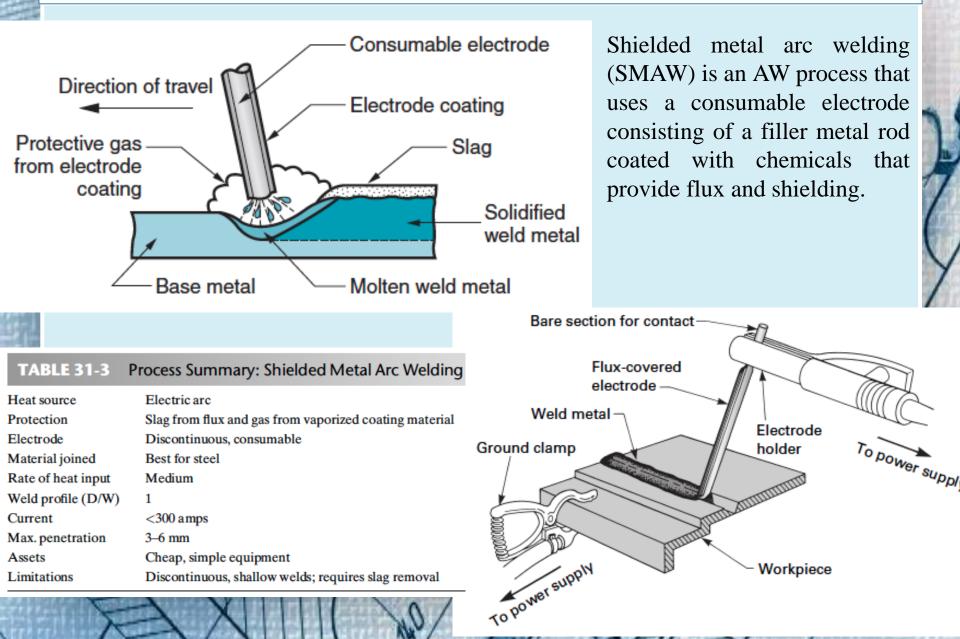


Arc Welding (AW)

- Arc welding (AW) is a fusion-welding process in which coalescence of the metals is achieved by the heat of an electric arc between an electrode and the work. The same basic process is also used in arc cutting.
- An electric arc is a discharge of electric current across a gap in a circuit. It is sustained by the presence of a thermally ionized column of gas (called a plasma) through which current flows.
- To initiate the arc in an AW process, the electrode is brought into contact with the work and then quickly separated from it by a short distance.



Shielded metal arc welding (SMAW)



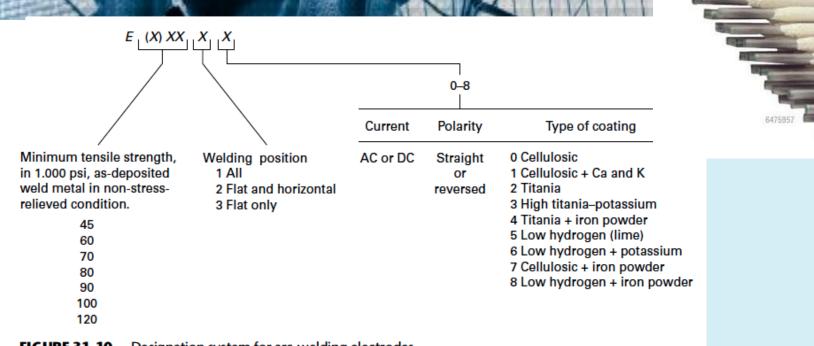


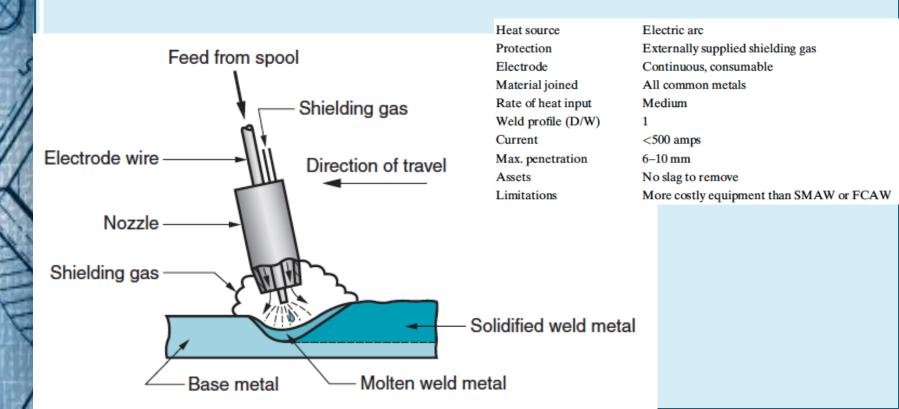
FIGURE 31-10 Designation system for arc-welding electrodes.



Gas metal arc welding (GMAW)

Gas metal arc welding (GMAW) is an AW process in which the electrode is a consumable bare metal wire, and shielding is accomplished by flooding the arc with a gas. The bare wire is fed continuously and automatically from a spool through the welding gun.

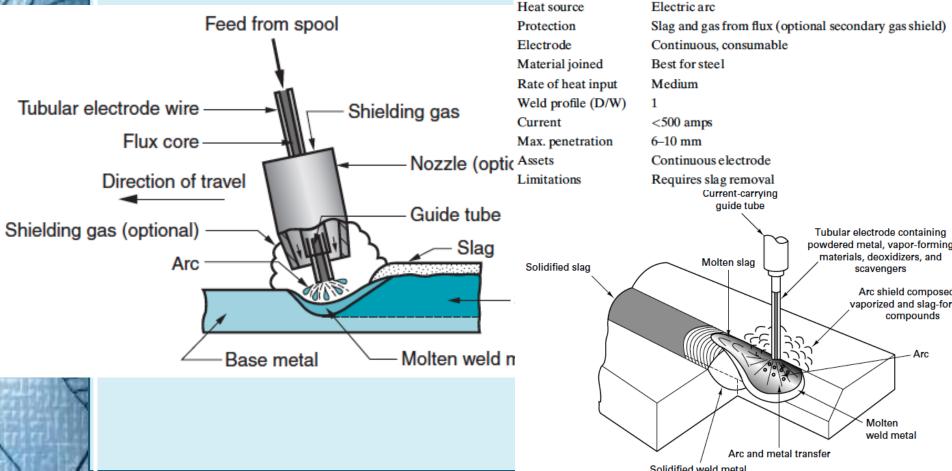
Selection of gases (and mixtures of gases) depends on the metal being welded, as well as other factors. Inert gases are used for welding aluminum alloys and stainless steels, while CO2 is commonly used for welding low and medium carbon steel. (MIG/MAG)





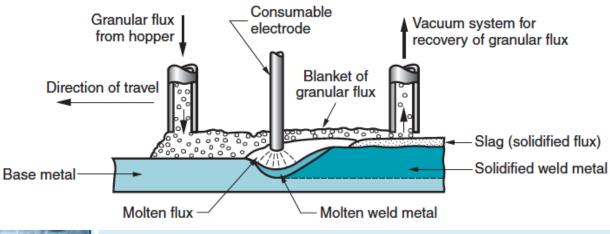
Flux-cored arc welding (FCAW)

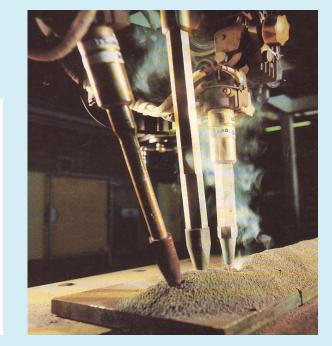
Flux-cored arc welding (FCAW) is an arc-welding process in which the electrode is a continuous consumable tubing that contains flux and other ingredients in its core. Other ingredients may include deoxidizers and alloying elements. The tubular flux-cored "wire" is flexible and can therefore be supplied in the form of coils to be continuously fed through the arc-welding gun.

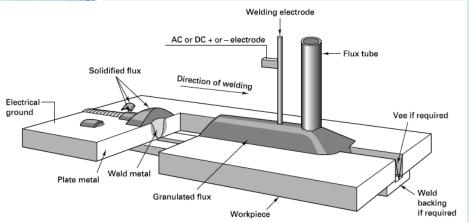




Submerged Arc Welding (SAW)





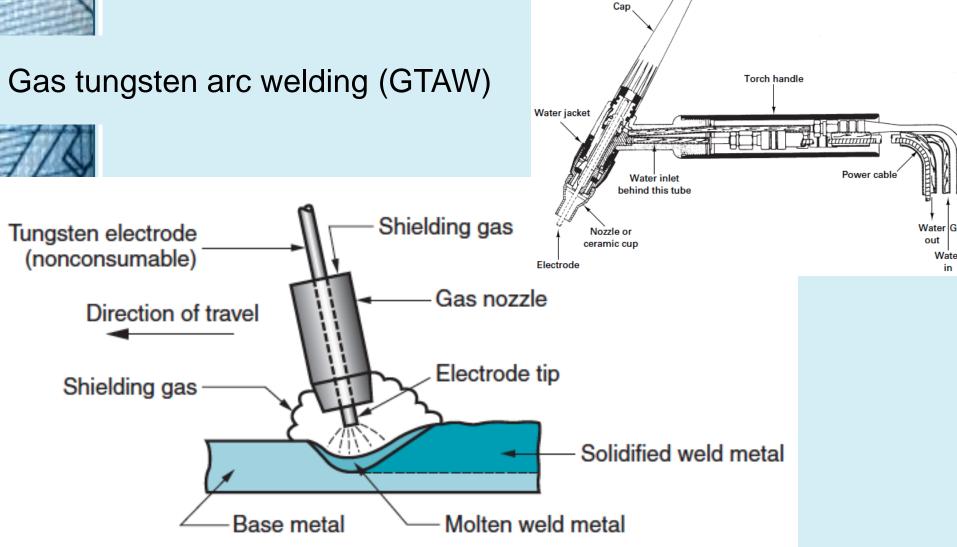


W

It is an arc-welding process that uses a continuous, consumable bare wire electrode, and arc shielding is provided by a cover of granular flux.

The electrode wire is fed automatically from a coil into the arc. The flux is introduced into the joint slightly ahead of the weld arc by gravity from a hopper.





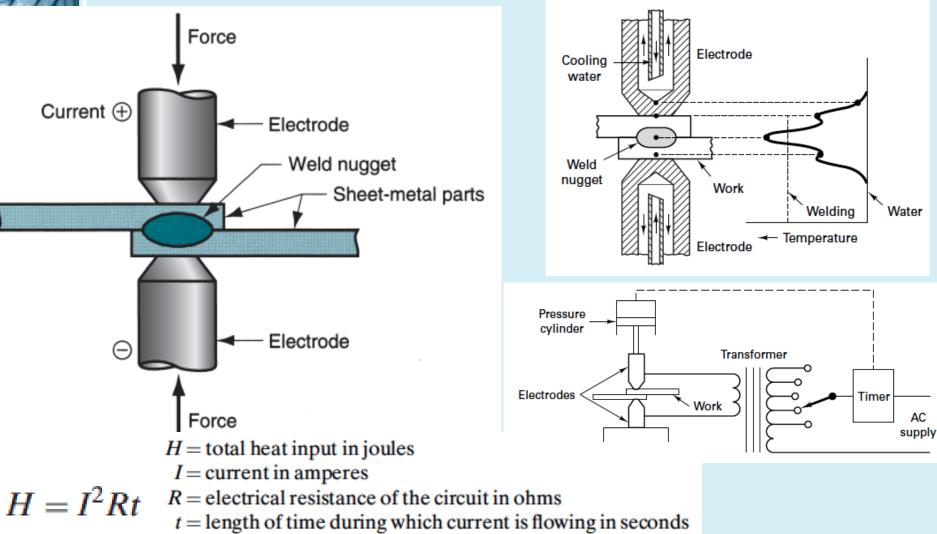


This process uses a nonconsumable tungsten electrode and an inert gas for arc shielding. The term TIG welding (tungsten inert gas welding) is often applied to this process.



Resistence Welding

Resistance welding (RW) is a group of fusion-welding processes that uses a combination of heat and pressure to accomplish coalescence, the heat being generated by electrical resistance to current flow at the junction to be welded.



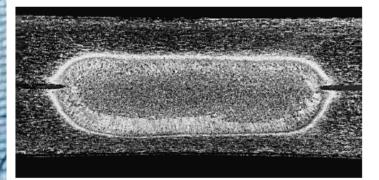


FIGURE 32-5 A spot-weld nugget between two sheets of 1.3-mm (0.05-in.) aluminum alloy. The nugget is not symmetrical because the radius of the upper electrode was greater than that of the lower electrode. (Courtesy Locheed Martin Corporation, Bethesda, MD)

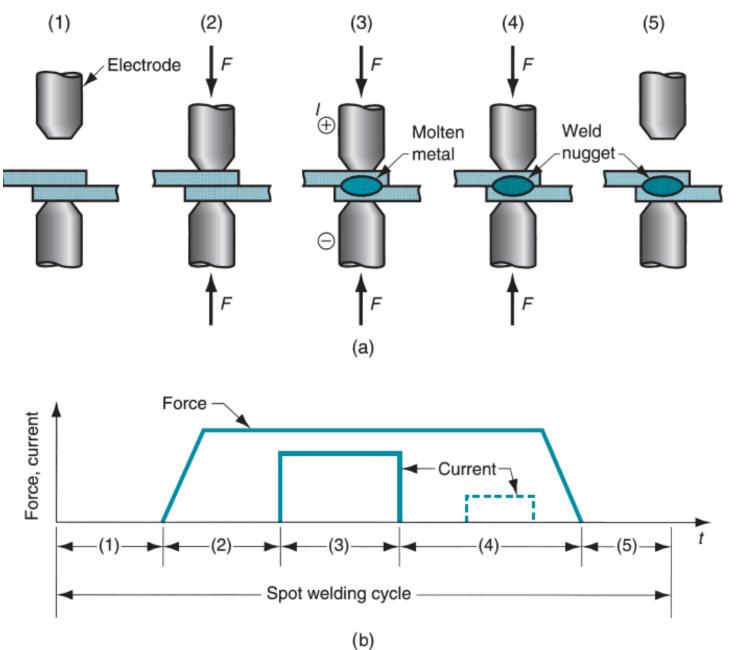




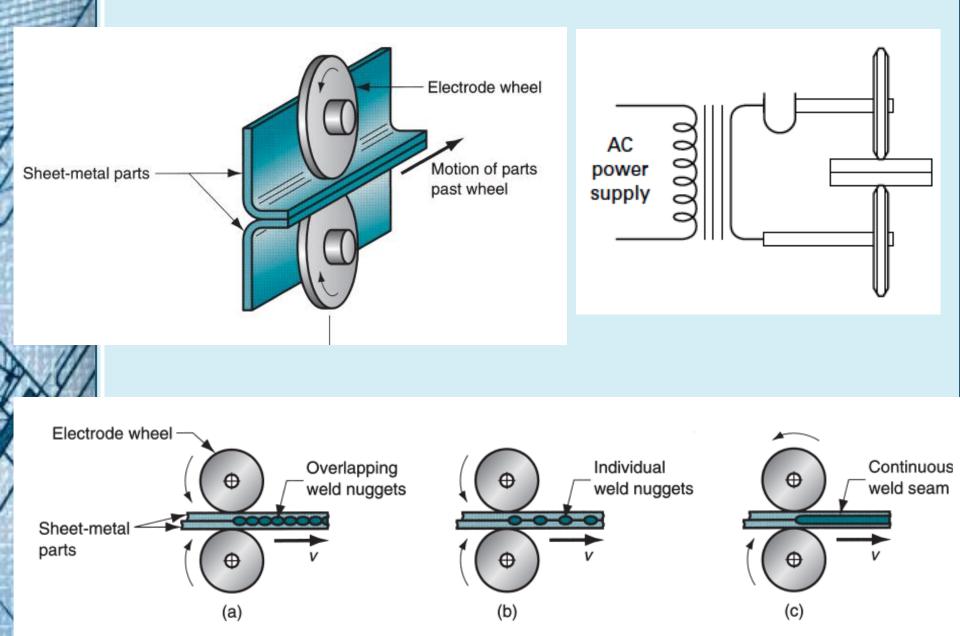
TABLE 32-1 Metal Combinations That Can Be Spot Welded

Metal	Aluminum	Brass	Copper	Galvanized Iron	Iron (Wrought)	Monel	Nichrome	Nickel	Nickel Silver	Steel	Tin Plate	Zinc
Aluminum	x										x	x
Brass		x	x	x	x	x	x	x	x	x	x	x
Copper		х	x	x	х	х	x	x	x	x	x	х
Galvanized Iron		x	x	x	x	x	x	x	x	x	x	
Iron (Wrought)		x	x	x	x	x	x	x	x	x	x	
Monel		х	x	x	х	х	x	x	x	x	x	
Nichrome		x	x	x	х	х	x	x	x	x	x	
Nickel		x	x	x	x	x	x	x	x	x	x	
Nickel Silver		x	x	x	x	x	x	x	x	x	x	
Steel		x	x	x	x	x	x	x	x	x	x	
Tin Plate	x	x	x	x	x	x	x	x	x	x		
Zinc	x	x	x									x

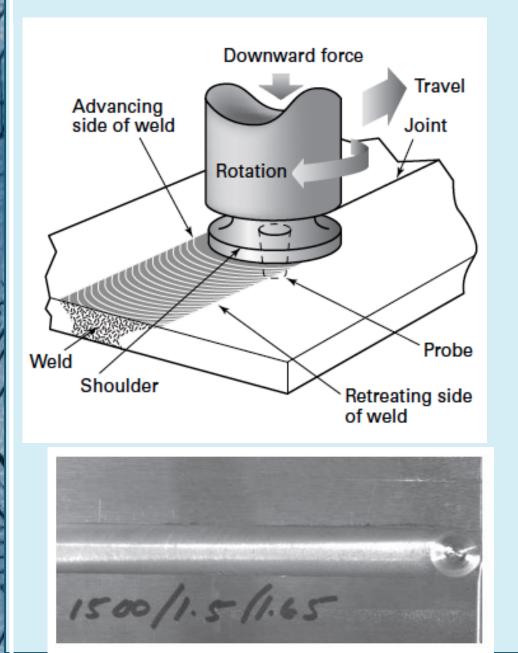
FIGURE 23.11 (a) Steps in a spot-welding cycle, and (b) plot of squeezing force and current during cycle. The sequence is: (1) parts inserted between open electrodes, (2) electrodes close and force is applied, (3) weld timecurrent is switched on, (4) current is turned off but force is maintained or increased (a reduced current is sometimes applied near the end of this step for stress relief in the weld region), and (5) electrodes are opened, and the welded assembly is removed. (Credit: Fundamentals of Modern Manufacturing, 4th Edition by Mikell P. Groover, 2010. Reprinted with permission of John Wiley & Sons, Inc.)

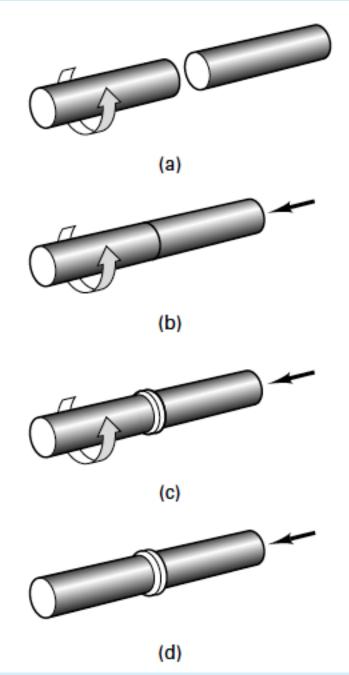


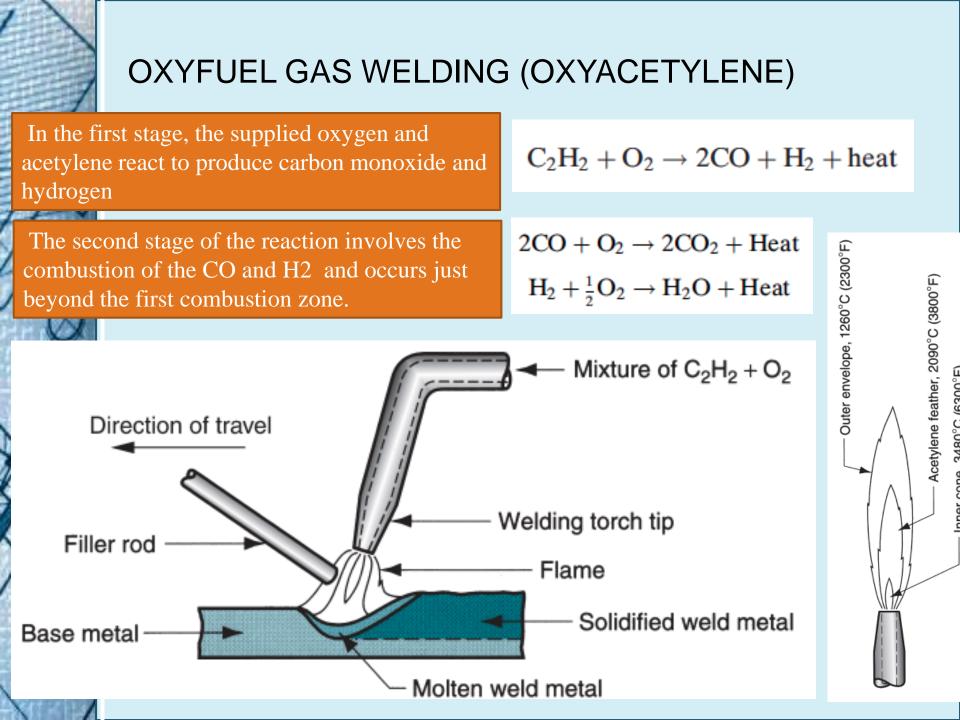
Resistance Seam Welding

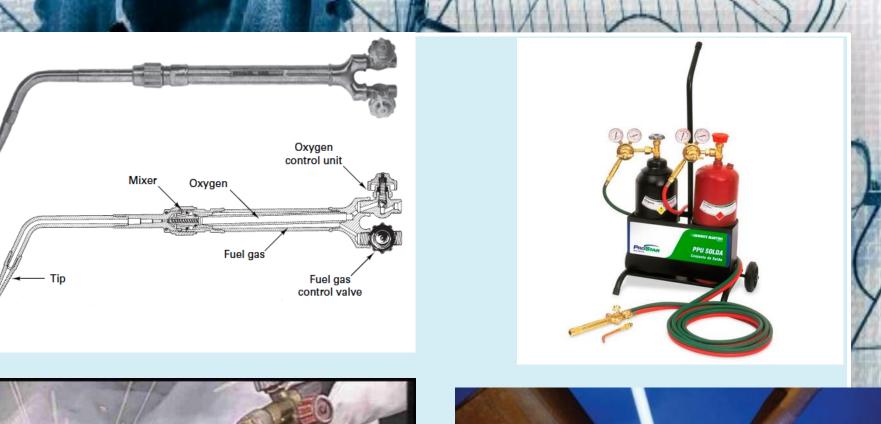


Friction Stir Welding

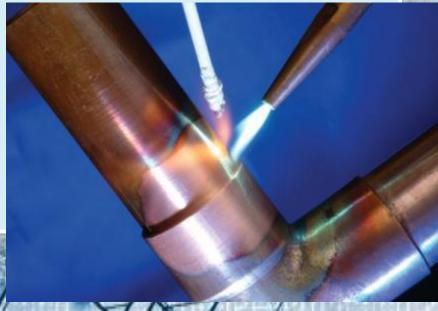








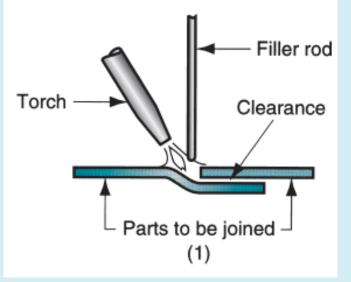




Brazing

metal.

- Brazing is a joining process in which a filler metal is melted and distributed by capillary action between the faying surfaces of the metal parts being joined. No melting of the base metals occurs in brazing; only the filler melts.
- In brazing, the filler metal (also called the brazing metal) has a melting temperature (liquidus) that is above 450C (840F) but below the melting point (solidus) of the base



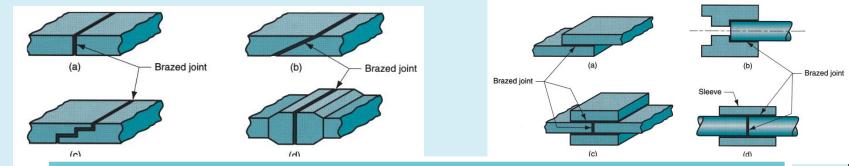
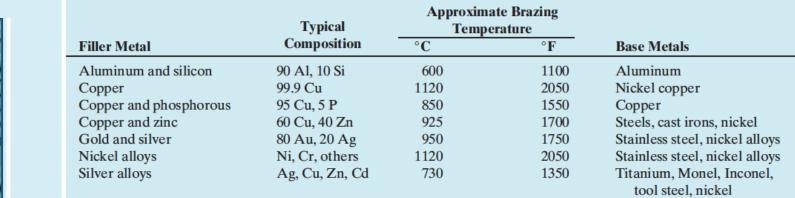


TABLE 24.1 Common filler metals used in brazing and the base metals on which they are used.





Soldering

Soldering is similar to brazing and can be defined as a joining process in which a filler metal with melting point (liquidus) not exceeding 450C (840F) is melted and distributed by capillary action between the faying surfaces of the metal parts being joined.

As inbrazing, no melting of the base metals occurs, but the filler metal wets and combines with the base metal to form a metallurgical bond.

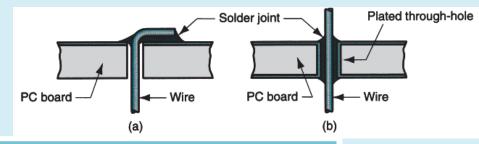
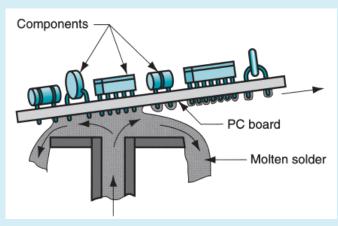


TABLE 24.2Some common solder alloy compositions with their meltingtemperatures and applications.

	Approximate	Approximate Melting Temperature		
Filler Metal	Composition	°C	°F	Principal Applications
Lead-silver	96 Pb, 4 Ag	305	580	Elevated temperature joints
Tin-antimony	95 Sn, 5 Sb	238	460	Plumbing & heating
Tin-lead	63 Sn, 37 Pb	183 ^a	361 ^a	Electrical/electronics
	60 Sn, 40 Pb	188	370	Electrical/electronics
	50 Sn, 50 Pb	199	390	General purpose
	40 Sn, 60 Pb	207	405	Automobile radiators
Tin-silver	96 Sn, 4 Ag	221	430	Food containers
Tin-zinc	91 Sn, 9 Zn	199	390	Aluminum joining
Tin-silver-copper	95.5 Sn, 3.9	217	423	Electronics: surface
	Ag, 0.6 Cu			mount technology



× ×

Adhesive Bonding

- Adhesive bonding is a joining process in which a filler material is used to hold two (or more) closely spaced parts together by surface attachment. The filler material that binds the parts together is the adhesive. It is a nonmetallic substance—usually a polymer.
- The parts being joined are called adherends. Adhesives of greatest interest in engineering are structural adhesives, which are capable of forming strong, permanent joints between strong, rigid adherends.

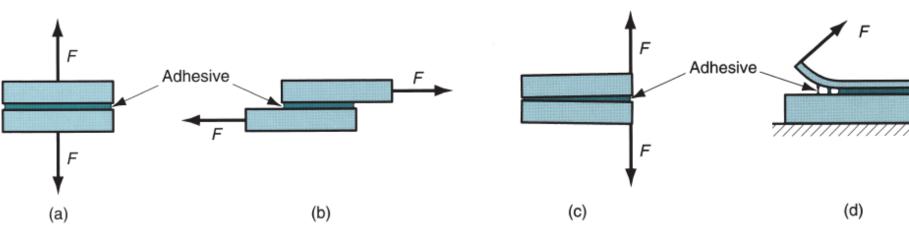


FIGURE 24.9 Types of stresses that must be considered in adhesive bonded joints: (a) tension, (b) shear, (c) cleavage, and (d) peeling. (Credit: *Fundamentals of Modern Manufacturing*, 4th Edition by Mikell P. Groover, 2010. Reprinted with permission of John Wiley & Sons, Inc.)

Adhesive Types

- 1. natural, derived from natural sources (e.g., plants and animals), including gums, starch, dextrin, soy flour, and collagen. This category of adhesive is generally limited to low-stress applications, such as cardboard cartons, furniture, and bookbinding; or where large surface areas are involved (e.g., plywood).
- 2. Inorganic adhesives are based principally on sodium silicate and magnesium oxychloride. Although relatively low in cost, they are also low in strength— a serious limitation in a structural adhesive.
- 3. Synthetic adhesives constitute the most important category in manufacturing. They include a variety of thermoplastic and thermosetting polymers. They are cured by various mechanisms, such as

(1) mixing a catalyst or reactive ingredient with the polymer immediately prior to applying,

(2) heating to initiate the chemical reaction,

(3) Radiation curing, such as ultraviolet light, and

(4) curing by evaporation of water from the liquid or paste adhesive.

In addition, some synthetic adhesives are applied as films or as pressuresensitive coatings on the surface of one of the adherends.

Joining of Plastics

Because the thermoplastics soften and melt at such low temperatures, the heat required to weld these materials is significantly less than that required in the welding of metals. The processes used to weld plastics can be divided into two groups:

- those that utilize mechanical movement and friction to generate heat, such as ultrasonic weld-spin welding, and vibration welding,
- (2) those that involve external heat sources, such as hot-plate welding, hot-gas welding, and resistive and inductive implant welding.

In both groups, it is important to control the rate of heating.

Plastics have low thermal conductivity, and it is easy to induce burning, charring or other material degradation before softening has occurred to the desired depth.

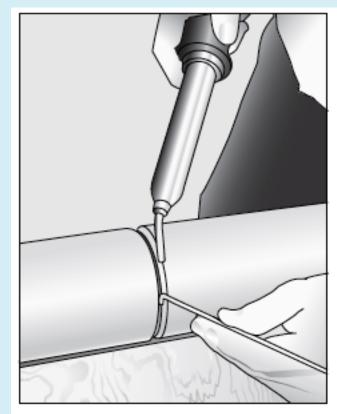


FIGURE 34-9 Using a hot-gas torch to make a weld in plastic pipe.



Joining of Ceramics

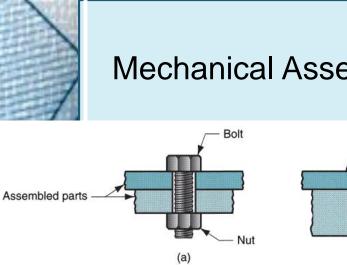
When we consider joining operations, the unique properties of ceramics once again introduce fabrication limitations.

Brittle ceramics cannot be joined by fusion welding or deformation bonding, and threaded assemblies should be avoided whenever possible.

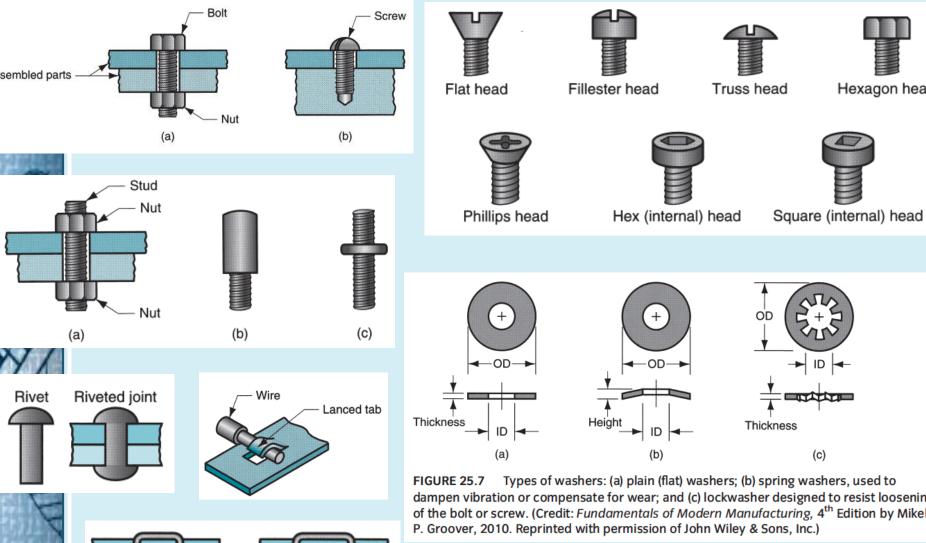
Therefore, most joining utilizes some form of adhesive bonding, brazing, diffusion bonding, or special cements.

Even with these methods, the stresses that develop on the surfaces can lead to premature failure.

As a result, most ceramic products are designed to be monolithic (single-piece) structures rather than multipart assemblies.



Mechanical Assembly





Mechanical Assembly and Quality Control

• Next Class